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temperature thermal insulation for heat engines and turbines. It is produced by the following reaction. $3\text{Si}(s) + 2\text{N}_2(g) \rightarrow \text{Si}_3\text{N}_4(s)$
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These problems are provided for each of the chapters for which additional mathematical problems would be beneficial. Most chapters contain 10-25 supplemental problems. You might use them as assessments or assign them for homework. Complete solutions can be found at the back of the Supplemental Problemsbook. To the Teacher

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the answer. 10 19 105 10 14; the answer will be about 20 10 14, or 2 10 13. c. Calculate your answer. Check it against your estimate from part b. 1.7 10 13 kg m/s² d. Justify the number of significant digits in your answer. The least-precise value is 4.5 T,

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with 2 significant digits, so the answer is rounded to 2 significant digits. 16.

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AnalysisData Analysis 1. A sample of aluminum is placed in a
25-mL graduated cylinder containing 10.0 mL of water. The level
of water rises to 18.0 mL. Aluminum has a density of 2.7 g/mL.
Calculate the mass of the sample. 2.

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