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Scott is for the one-semester General, Organic and Biological Chemistry course. The authors designed this textbook from the ground up to meet the needs of a one-semester course.

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In chemistry, a solution is a special type

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of homogeneous mixture composed of two or more substances. In such a mixture, a solute is a substance dissolved in another substance, known as a solvent. The mixing process of a solution happens at a scale where the effects of chemical polarity are involved, resulting in interactions that are specific to solvation.

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Introduction. A simple demonstration of

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how a precipitate is evidence of a chemical reaction taking place is performed by mixing solutions of calcium chloride and sodium carbonate to form the precipitate calcium carbonate ( $\text{CaCO}_3$ ).

$$\text{CaCl}_2 (\text{aq}) + \text{Na}_2\text{CO}_3 (\text{aq}) \rightarrow \text{CaCO}_3 (\text{s}) + 2\text{NaCl}(\text{aq}).$$

To Conduct Demonstration

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