

Chemquest 21 Transition Metals Answers

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ChemQuest 21 Name: ____ Date: ____ Hour: ____ Information: Charges of Some Transition Elements So far you have learned that you can predict the charge that an ion will have based on its location on the periodic table. However, the transition elements are not easy to predict. A few common

Information : Charges of Some Transition Elements

Many transition elements can have more than one charge when they become an ion. Copper ions, for example, can be Cu^+ or Cu^{2+} . As another example, iron ions are sometimes Fe^{2+} and sometimes Fe^{3+} . Critical Thinking Questions. 2. Copper and iron are in the "d block" and so you need to calculate their charge by comparing what bonds to them.

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70 ChemQuest 21 Name: ____ Date: ____ Hour: ____ Introduction Questions 1. Draw the Lewis structures for the following molecules. ... Is your answer to question 3 consistent with the data in the table Hopefully you see that triple bonds require more energy to break—they are harder to

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break than other bonds. 5. What is the relationship ...

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