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$3\text{Zn}(\text{s}) + 8\text{HNO}_3(\text{aq}) \rightarrow 3\text{Zn}(\text{NO}_3)_2(\text{aq}) + 2\text{NO}(\text{g}) + 4\text{H}_2\text{O}(\text{l})$ The concentration of copper(II) ions in the resulting solution is then determined from a . calibration curve, which is plotted by measuring the light absorbance of standard solutions.

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Calibration curve. 0 0 00 040 00 080 100
10 000 0010 001 000 00 0030

Concentration of Cu. 2 + ion / mol dm –
3

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The maximum mark for this examination
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Option A — Modern analytical chemistry 1 - 3
Option B — Human biochemistry 4 - 6
Option C — Chemistry in industry and technology 7 - 10
Option D — Medicines and drugs 11 - 13
Option E — Environmental chemistry 14 - 16

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CH₃ CH₃ CH₃ Cholesteryl benzoate (a)
Identify the structural feature of
cholesteryl benzoate which makes it
suitable for use as a liquid crystal. [1].....
(b) Suggest the essential feature a liquid-
crystal molecule must have so that the
display can be turned “on” and “off”.
[1]..... (c) Outline the principles of a
liquid-crystal display (LCD) device. [3]

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marks is 30.

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Option C — Chemistry in industry and
technology 7. (a) (i) + 8 3 / 2 2 3 + ;
Accept +2.7 but not +3. +2 and +3 /
contains two (or more) iron ions with
different oxidation states / contains

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Fe²⁺ and Fe³⁺; [2] Accept II and III oxidation number notation for oxidation states but not 2+ and 3+ unless ions are referred to explicitly.

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- 3 - Subject Details: Chemistry SL Paper
3 Markscheme Mark Allocation

Candidates are required to answer ALL

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questions in Section A [15 marks] and all questions from ONE option in Section B [20 marks]. Maximum total = [35 marks].

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